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Context-Dependent Photodimerization in Isolated Thymine–Thymine Steps in DNA

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Dimerization of thymine has long been recognized as a leading cause of cellular photodamage by ultraviolet light.^{1,2} A revival of interest in T-T dimerization has been driven by advances in fast spectroscopy,^{3,4} studies of ground-state conformational dynamics^{5,6} and excited-state potential energy surfaces,⁷ and synthetic nucleotide chemistry.^{8,9} The cis-syn cyclobutane dimer formed in a [2+2]cycloaddition reaction between adjacent thymines is the major product of irradiation of duplex DNA, accompanied in some cases by lesser amounts of the 6-4 adduct, which is thought to arise from a Paterno-Büchi cycloaddition followed by thermal ring opening of the oxetane intermediate (Chart 1a).¹⁰ Both the efficiency of product formation and the product ratio in duplex DNA are dependent upon the base sequence, lower yields being observed for flanking purines vs pyrimidines.^{11,12} Dimerization efficiency has also been related to conformational flexibility.^{2,12,13} Recently attention has focused on the importance of ground-state conformational populations in determining T-T dimerization efficiency.^{4,5}

We report here an investigation of T–T photodimerization in a family of minihairpins possessing only A–T base pairs and a single T–T step connected by an alkane linker (C12, Chart 1b). Minihairpins having short, well-defined base pair sequences with chromophores located at one or both ends have been employed in our studies of electron transfer, energy transfer, and exciton coupling in DNA.¹⁴ The use of hairpins with tetranucleoside loops for the study of the base-sequence dependent formation of UV-induced lesions has been reported by Carell et al.⁹ However, the context dependence of the photodimerization of hairpins possessing a single T–T step as a function of its location has not been explored. We find that both the efficiency and products of photodimerization are remarkably dependent upon the location of the T–T step within a duplex domain and in conjugates possessing a T bulge or overhang.

Oligonucleotide conjugates 1-10 and single strand sequences 11a,b (Chart 1) were prepared using standard phosphoramidite chemistry, purified using reversed-phase HPLC, and characterized by MALDI-TOF mass spectrometry (Table S1). Conjugates 1-10 exhibit UV and CD spectra (Figures S1, S2) characteristic of poly(A-T) duplexes or hairpins.¹⁵ Conjugates 1-6 have melting temperature of 45.5-49.5 °C (Table S2) obtained from thermal dissociation profiles in 10 mM phosphate buffer (pH 7.2) containing 100 mM sodium chloride (standard buffer). Conjugate 7 (single T bulge) has a lower $T_{\rm M}$ (40.1 °C), and 8 and 9 (single and double T overhang) and 10 (continuous A-tract) have higher $T_{\rm M}$ values (Table S2). Solutions containing ca. $1-1.2 \,\mu$ M hairpin in standard buffer were irradiated with monochromatic 280 nm light at 10 °C in 1 cm path length quartz cuvettes. The progress of irradiation was monitored at 260 nm by high temperature HPLC with a column temperature of 60 °C, a procedure similar to that employed by Barton and co-workers in their studies of thymine dimer repair.¹⁶ The use of high temperature HPLC to achieve thermal rather than chemical denaturation permits analysis of small aliquots of irradiated solutions with excellent resolution.

Chart 1^a







Figure 1. HPLC traces a-e for irradiation of hairpins (a) **6** and (b) **7** at 280 nm with increasing irradiation times 0, 10, 20, 30, and 40 min, respectively. Solutions were analyzed by high temperature reversed-phase HPLC on a Microsorb MV C₁₈ reversed-phase column maintained at 60 °C using a gradient of 20 mM ammonium acetate containing 7% acetonitrile to 12% acetonitrile.

Representative HPLC traces obtained at 10 min irradiation intervals for hairpins 6 and 7 are shown in Figure 1. The growth of a single dominant product peak at a shorter retention time than the starting material is observed for 2-5, 7-9, and the single strand 11a (Figure 1b). However, in the case of 6 two product peaks are formed in a ca. 45:55 ratio (Figure 1a). No new product is observed upon irradiation of hairpin 1 or single strand 11b which lack a TT step, and multiple product peaks are observed for hairpin 10 which has a polyT sequence. All of the products obtained from hairpins 2-9 have the same mass as the starting material. The isolated product from hairpin 5 has a value of $T_{\rm M} = 30$ °C, significantly lower than that of 5 (47.5 °C), as expected for the formation of a photolesion.¹⁷ All of the product peaks except for the minor product from 6 have UV spectra similar to those of the starting materials (Figure S3a) and undergo reversion to starting materials upon irradiation at 240¹⁸ or 254 nm,¹⁹ as expected for 2+2 thymine dimers. The minor product from 6 is tentatively assigned to the 6-4 adduct based on its weak absorption band at 320 nm (Figure S3b) and stability upon irradiation at 240 nm.²⁰



Figure 2. Relative percentage dimer formation of hairpin sequences 1-8 as determined using HPLC with increasing time of irradiation at 280 nm. Data for 6 are the sum of both products.

The time-dependent conversion of hairpins 1-8 to photoproducts upon irradiation is shown in Figure 2. The linearity of these plots to high conversions is a consequence of the low absorbance of the photoproducts at the 280 nm irradiation wavelength. Hairpins 3-5 which have a TT step within the interior of their A-T base pair domains display similar conversions of ca. 30% after 45 min of irradiation. Using their conversions as a basis for comparison, significantly lower conversions are observed for hairpins 2, 8, and 9 (13-14% conversion) and higher conversions for hairpins 6 (60%) total conversion), 7 (53% conversion), and 10 (95% total conversion). Single strand 11a was irradiated in a 1:1 mixture of the noncomplementary strand 11b to provide a ratio of reactive and nonreactive T and A chromophores similar to that for the hairpins. The conversion observed for **11a** is 14%, similar to that for hairpins 2, 8, and 9.

Our results for hairpins 1-5 and single strands 11a,b are consistent with previous reports that the 2+2 dimer is the major product obtained from TT steps in either duplex and single strand DNA but is not formed from alternating AT sequences.9 Unprecedented are the observations of (a) identical reactivity for the interior TT steps in 3-5, (b) diminished dimerization efficiency for TT steps at the 5'-3' hairpin terminus of 2 and the single and double T and TT overhangs in 8 and 9, as well as in the single strand 11a, (c) enhanced 2+2 dimerization at the site of a single T bulge in 7, and (d) formation of both 2+2 and 6-4 products adjacent to the alkane linker in 6. These observations can be understood in terms of the topochemical principles of Fischer for solid state photodimerization²¹ as applied by Kohler and others to T-T dimerization in DNA.4 That is, dimerization occurs only in a subset of groundstate conformations in which the adjacent thymines have their chromophores properly aligned so that bonding can compete with ultrafast nonradiative decay or energy transfer.

Diminished reactivity in hairpins 2, 8, and 9 suggests that TT steps at the hairpin terminus and the overhangs in 8 and 9 (like the single strand 11a) are less likely to adopt reactive conformations than are TT steps in the B-DNA interior of 3-5. Solution NMR studies of duplexes having terminal A-T base pairs or T overhangs display line broadening indicative of a high degree of conformational mobility or end-fraying.²² Thus conformational mobility is not a good predictor of high dimerization efficiency. The enhanced reactivity of hairpin 7 is indicative of a "stacked-in" rather than a "looped-out" conformation for the "extra" T.²³ The formation of a 6-4 adduct from hairpin 6 finds analogy in reports by Clivio and co-workers for TpT dinucleotides having C3'-endo sugar conformations.⁸ The enhanced reactivity of **6** and **7** suggests that the presence of reactive conformations rather than flexibility is the best indicator of T-T dimerization efficiency, as suggested by recent molecular dynamics simulations for single strand poly(T).⁵ Solution NMR studies and molecular dynamic simulations of the C12-linked hairpins 6 and 7 are in progress as are further studies of the contextdependent photodimerization in isolated TT steps.

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Supporting Information Available: MALDI-TOF analysis of the hairpin sequences 1-10 and the single strand sequences 11a and 11b, melting points of sequences 1-10, representative UV and circular dichoism spectra of sequence 1, change in absorption spectra of 2 and 6 with increasing irradiation time. This material is available free of charge via the Internet at http://www.pubs.acs.org.

References

- (a) Setlow, R. B. Science **1966**, *153*, 379–386. (b) Lamola, A. A.; Eisinger, J. Proc. Natl. Acad. Sci. U.S.A. **1968**, 59, 46–51. (c) Taylor, J. S.; Brockie, I. R.; O'Day, C. L. J. Am. Chem. Soc. **1987**, *109*, 6735–6742. (d) Park, H.; Zhang, K.; Ren, Y.; Nadji, S.; Sinha, N.; Taylor, J. S.; Kang, C. Proc. Natl. Acad. Sci. U.S.A. **2002**, *99*, 15965–15970.
 C. M. L. M. M. Marker, M. Marker, M. M. Ma
- (2) Cadet, J.; Vigny, P. In The Photochemistry of Nucleic Acids; Morrison, H., Ed.; Wiley: New York, 1990.
- (a) Crespo-Hernandez, C. E.; Cohen, B.; Kohler, B. Nature 2005, 436, 1141–1144. (b) Kwok, W. M.; Ma, C.; Phillips, D. L. J. Am. Chem. Soc. 2008, 130, 5131–5139. (c) Marguet, S.; Markovitsi, D. J. Am. Chem. Soc. 2005, 127, 5780–5781. (d) Samoylova, E.; Lippert, H.; Ullrich, S.; Hertel, I. V.; Radloff, W.; Schultz, T. J. Am. Chem. Soc. 2005, 127, 1782–1786. (e) Buchvarov, I.; Wang, Q.; Raytchev, M.; Trifonov, A.; Fiebig, T. Proc. Natl. Acad. Sci. U.S.A. 2007, 104, 4794–4797.
- Schreier, W. J.; Schrader, T. E.; Koller, F. O.; Gilch, P.; Crespo-Hernandez, C. E.; Swaminathan, V. N.; Carell, T.; Zinth, W.; Kohler, B. Science 2007, 315, 625-629
- (a) Law, Y. K.; Azadi, J.; Crespo-Hernandez, C. E.; Olmon, E.; Kohler, B. Biophys. J. 2008, 94, 3590–3600. (b) Johnson, A. T.; Wiest, O. J. Phys. Chem. B 2007, 111, 14398-14404.
- (a) Zhang, R. B.; Eriksson, L. A. J. Phys. Chem. B 2006, 110, 7556-7562. (b) Durbeej, B.; Eriksson, L. A. J. Photochem. Photobiol. A: Chem. 2002, 152, 95-101.
- (7) (a) Boggio-Pasqua, M.; Groenhof, G.; Schaefer, L. V.; Grubmueller, H.; (7) (a) Boggio-Fasqua, M., Groemor, C., Schaerer, L. V., Gruomener, H., Robb, M. A. J. Am. Chem. Soc. 2007, 129, 10996–10997. (b) Blancafort, L.; Migani, A. J. Am. Chem. Soc. 2007, 129, 14540–14541.
 (8) (a) Moriou, C.; Thomas, M.; Adeline, M. T.; Martin, M. T.; Chiaroni, A.; Pochet, S.; Fourrey, J. L.; Favre, A.; Clivio, P. J. Org. Chem. 2007, 72, View Context Science 2007, View Context Science 2007, 72, View Context Science 2007, View Context Science 2007, 72, View Context Science 2007, View Context Science 2007, 72, View Context Science 2007, View Context
- 43-50. (b) Desnous, C.; Babu, B. R.; Moriou, C.; Mayo, J. U. O.; Favre, A.; Wengel, J.; Clivio, P. J. Am. Chem. Soc. 2008, 130, 30–31.
 Kundu, L. M.; Linne, U.; Marahiel, M.; Carell, T. Chem.-Eur. J. 2004,
- 10, 5697-5705.
- (10) (a) Beukers, R.; Berends, W. Biochim. Biophys. Acta 1960, 41, 550–551.
 (b) Beukers, R.; Eker, A. P. M.; Lohman, P. H. M. DNA Repair 2008, 7, 530-543.
- (11) Bourre, F.; Renault, G.; Seawell, P. C.; Sarasin, A. Biochimie 1985, 67 (3-4), 293-299.
- (12) Becker, M. M.; Wang, Z. J. Mol. Biol. 1989, 210 (3), 429-438.
- (13) Tramer, Z.; Wierzchowski, K. L.; Shugar, D. Acta Biochim. Pol. 1969, 16, 83-107
- (14) (a) Lewis, F. D.; Liu, X.; Liu, J.; Miller, S. E.; Hayes, R. T.; Wasielewski, M. R. Nature 2000, 406, 51–53. (b) Lewis, F. D.; Zhang, L.; Zuo, X. J. Am. Chem. Soc. 2005, 127, 10002–10003. (c) Lewis, F. D.; Zhang, L.; Liu, X.; Zuo, X.; Tiede, D. M.; Long, H.; Schatz, G. C. J. Am. Chem. Soc. 2005, 127, 14445-14453
- (15) Lewis, F. D.; Wu, Y.; Liu, X. J. Am. Chem. Soc. 2002, 124, 12165-12173. (16) Dandliker, P. J.; Holmlin, E. K.; Barton, J. K. Science 1997, 275, 1465-1468

- (17) O'Neil, L. L.; Wiest, O. J. Phys. Chem. B 2008, 112, 4113–4122.
 (18) Deering, R. A.; Setlow, R. B. Biochim. Biophys. Acta 1963, 68, 526–534.
 (19) Holman, M. R.; Ito, T.; Rokita, S. E. J. Am. Chem. Soc. 2007, 129, 6–7. (20) Matsunaga, T.; Hieda, K.; Nikaido, T. Photochem. Photobiol. 1991, 54,
- 403-410 (21) Fischer, G. J.; Johns, H. E. Pyrimidine photodimers: In Photochemistry
- and Photobiology of Nucleic acids; Academic Press: New York, 1976. (22)Lane, A.; Martin, S. R.; Ebel, S.; Brown, T. Biochemistry 1992, 31, 12087-12095
- (23) Joshua-Tor, L.; Frolow, F.; Appella, E.; Hope, H.; Rabinovich, D.; Sussman, J. L. J. Mol. Biol. 1992, 225, 397–431.

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